

Mechanism of curcumin interaction and inclusion by two β -cyclodextrin rings: an MD study

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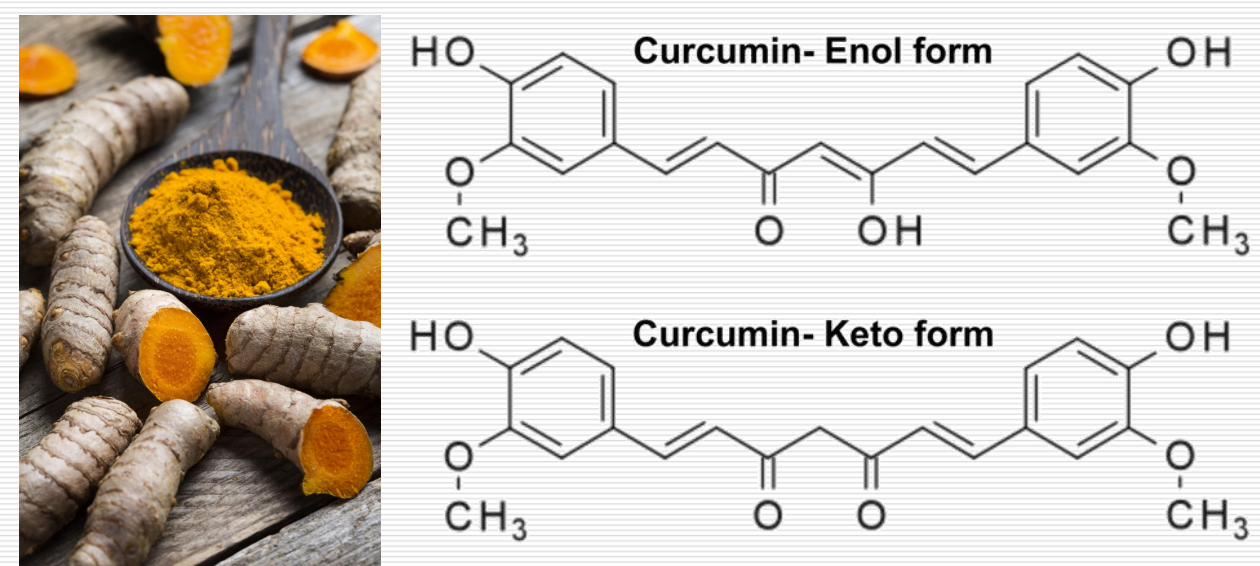


1. INTRODUCTION

Curcumin (CUR):

- hydrophobic polyphenol; turmeric-derived;
- broad pharmacological activity;

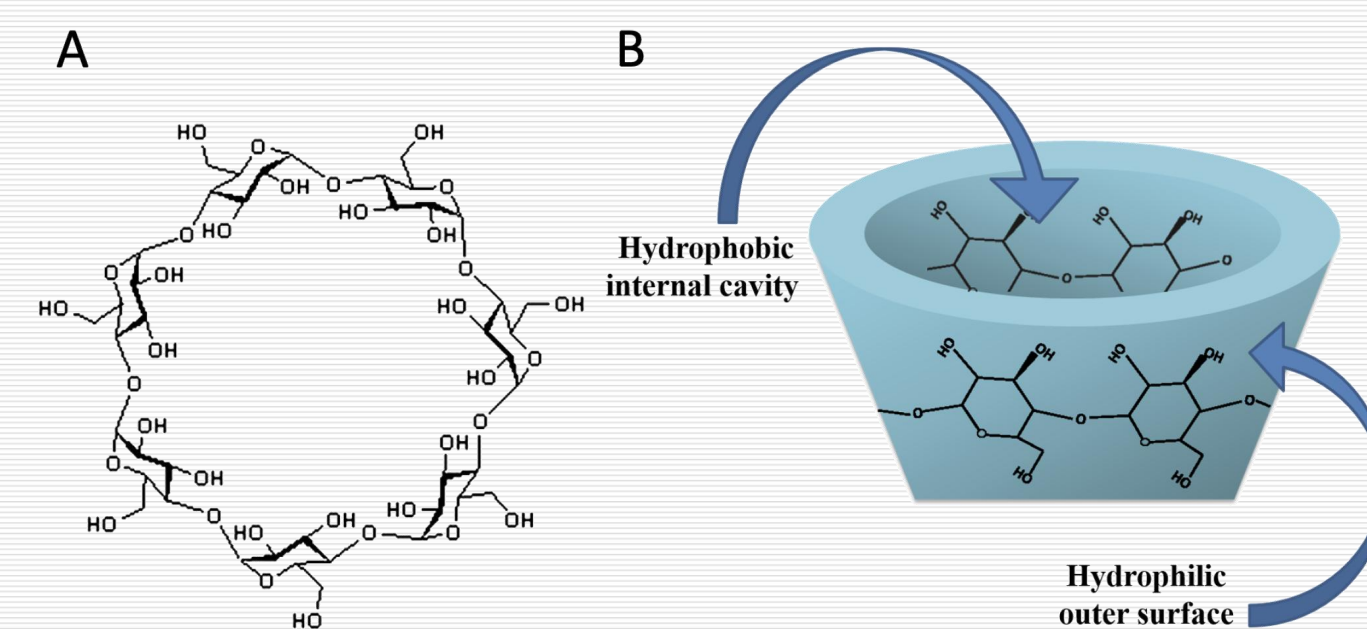
BUT poor solubility; low bioavailability



IUPAC	(1E,6E)-1,7-Bis(4-hydroxy-3-methoxyphenyl)-hepta-1,6-diene-3,5-dione
Chemical formula	C ₂₁ H ₂₀ O ₆
Solubility (mg/mL)	0.0004 at pH 7.3 [Chen et al, Int J Nanomedicine, 2020]

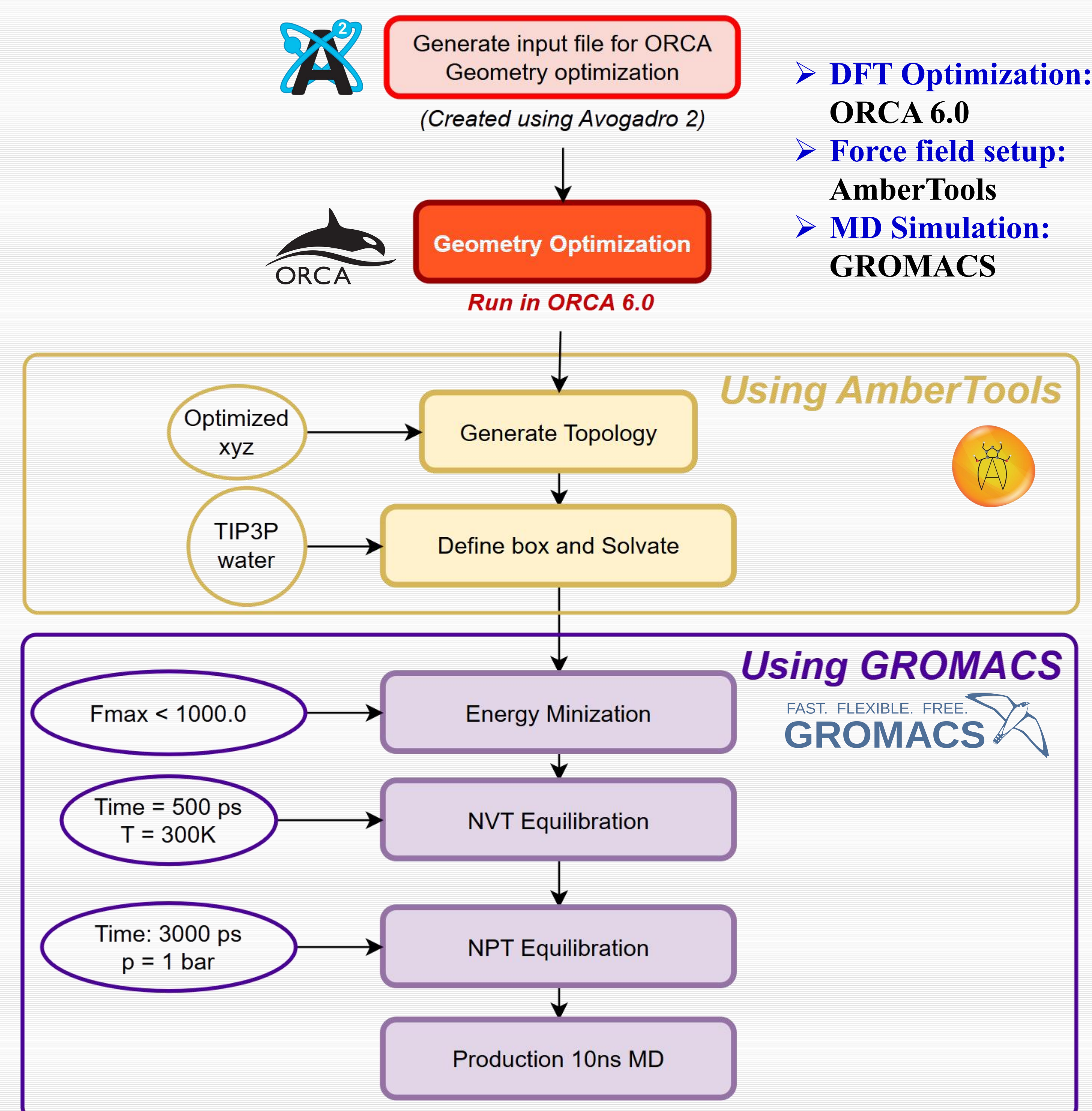
β -Cyclodextrin (β -CD):

- Hydrophobic cavity
 - Hydrophilic surface
- CUR stability \uparrow via β -CD inclusion



⇒ **Aim:** Investigate the stability, interaction and inclusion behavior of CUR-(β -CD)₂ by MD simulations.

2. METHODOLOGY



3.2. Root-Mean-Square Deviation (RMSD)

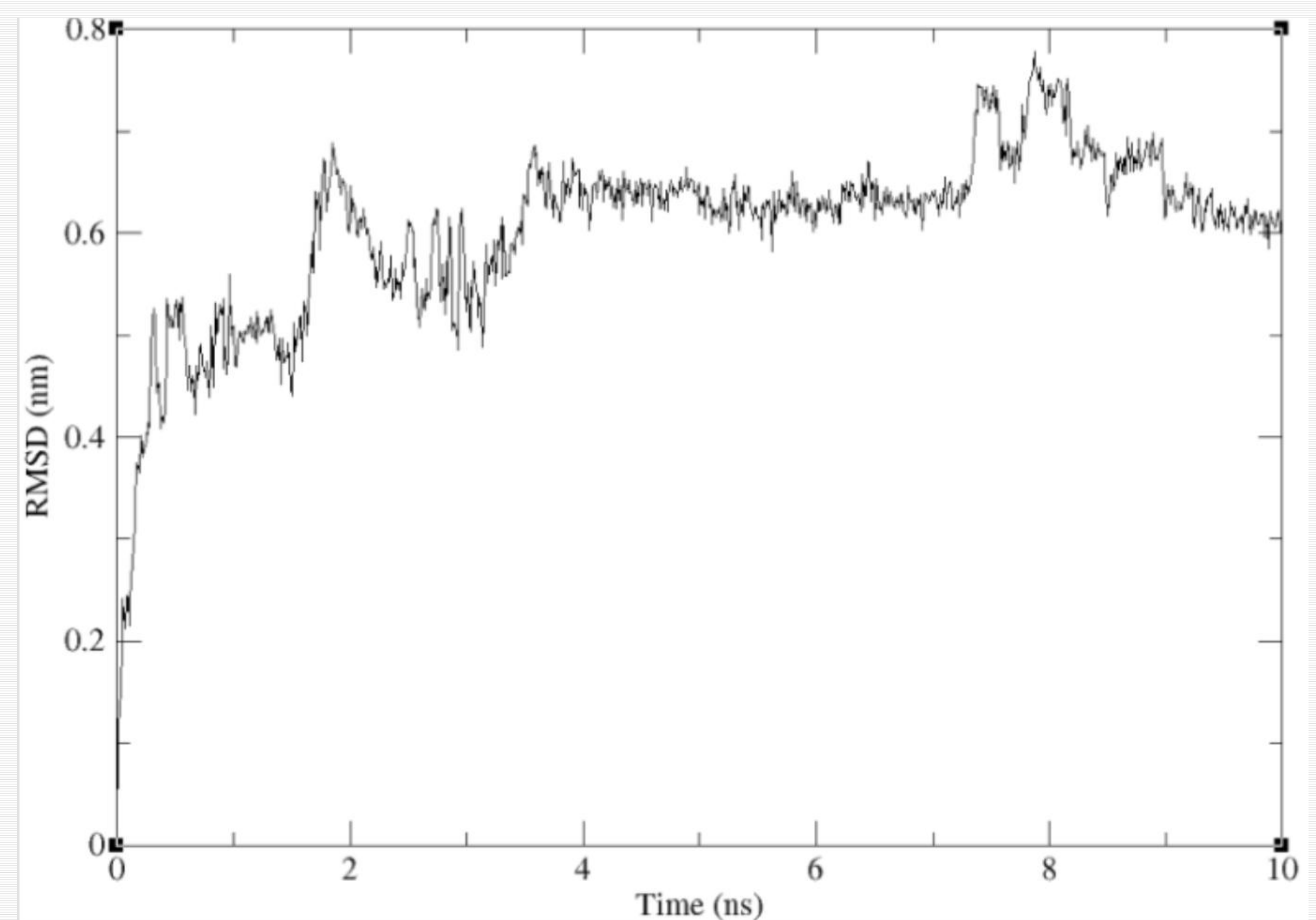


Fig. 3.2. RMSD of CUR-(β -CD)₂ over 10ns MD simulation.

3.3. Radius of Gyration

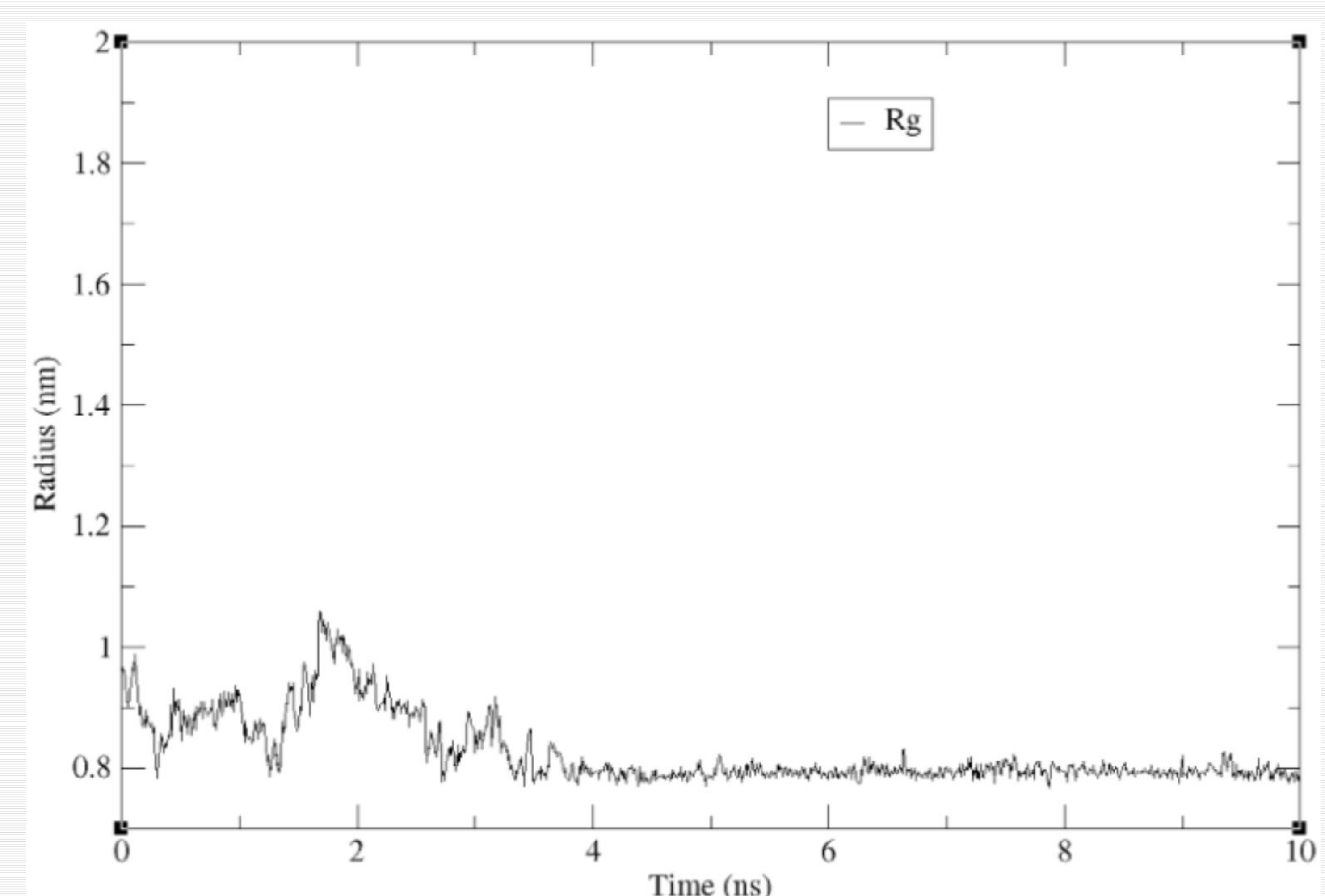


Fig. 3.3. Radius of Gyration of CUR-(β -CD)₂ over 10ns MD simulation.

➤ Equilibrium reached at ~8 ns; stable RMSD and Radius of Gyration.

3.4. Number of Hydrogen bonds

- Hydrogen bonds between CUR and β -CD: sparse but stable.
- β -CD-water H-bonds: remained abundant.

This confirmed host solvation stability.

3.5. Nonbonded Interaction Energies between CUR and β -CDs

- Short-range Lennard Jones \gg Short-range Coulombic.

⇒ van der Waals & hydrophobic forces are the main stabilizing factors.

3.6. Solvent-accessible surface area (SASA)

- SASA of CUR encapsulated by β -CDs significantly decreased compared with that of free CUR.

⇒ Tight and persistent inclusion.

3. RESULTS AND DISCUSSIONS

3.1. Structure of CUR-(β -CD)₂ complex

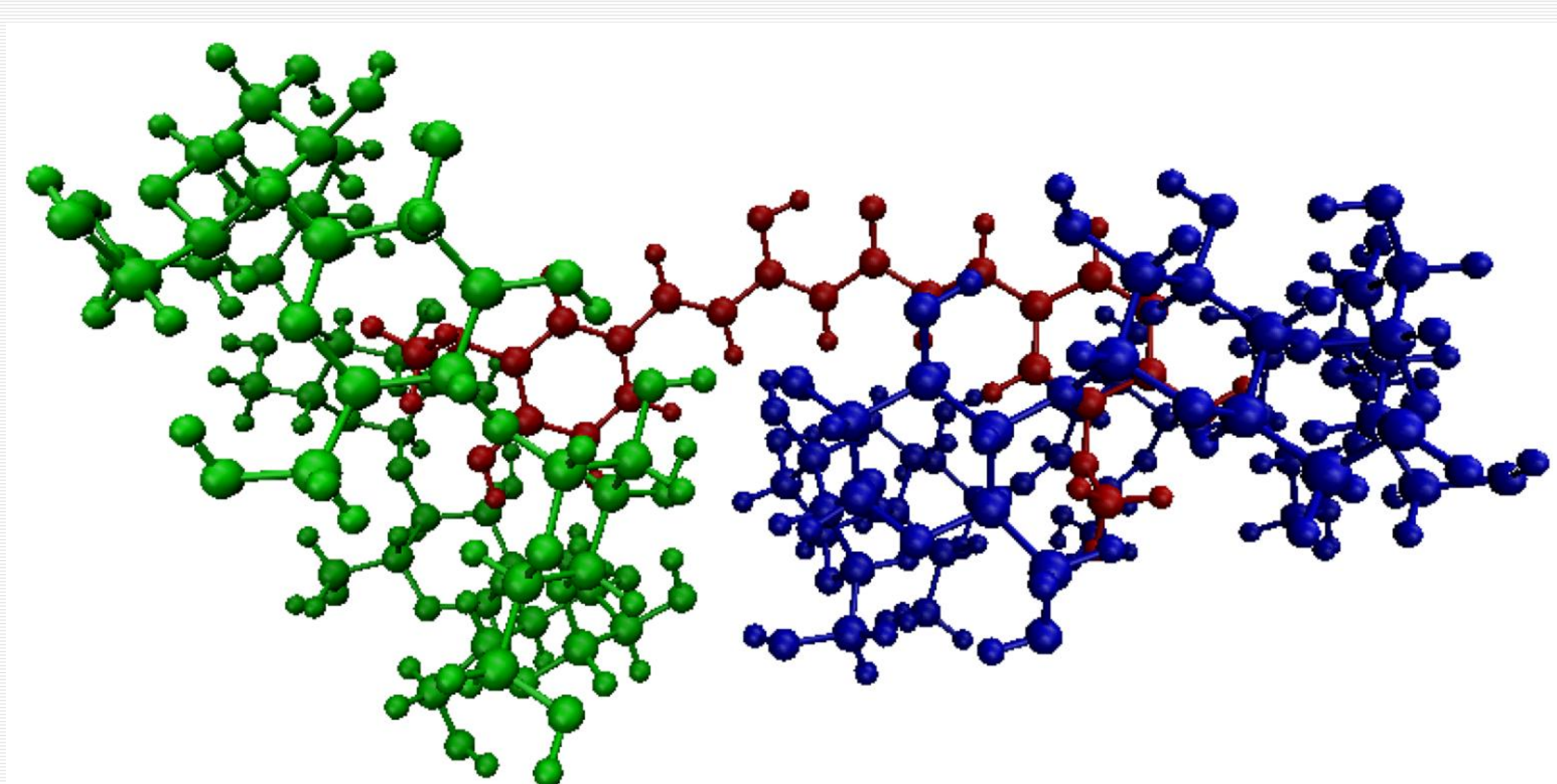


Fig. 3.1. Structure of CUR-(β -CD)₂ after geometry optimization: CUR (red) and β -CDs (green and blue).

4. CONCLUSION

- ❖ Two β -CD rings cooperatively encapsulate CUR into a compact capsule.
- ❖ Stability driven mainly by nonpolar interactions.
- ❖ Provides a molecular basis for future release-mechanism simulations and rational drug-carrier design.