

A practical computational protocol for photocatalytic reactions beyond ground-state DFT approximations

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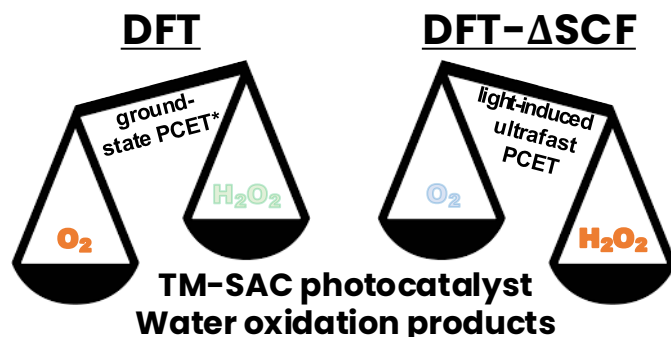
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Summary

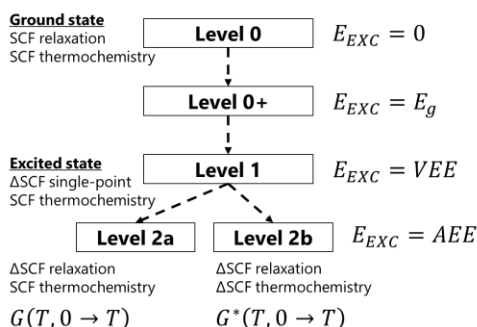
COMPUTATIONAL STUDIES OF HETEROGENEOUS PHOTOCATALYSTS TYPICALLY DISCUSS THE BAND LEVEL ALIGNMENT OBTAINED BY REGULAR DFT, WHICH DOES NOT CAPTURE THE PHYSICS OF LIGHT-DRIVEN PROCESSES. IN A NEW COMPUTATIONAL PROTOCOL, EXCITED STATES ARE EXPLICITLY CONSIDERED IN THE GIBBS FREE ENERGY DIAGRAMS. APPLIED TO PROTOTYPICAL REACTIONS ON A SINGLE-ATOM COCATALYST, THE PROTOCOL ALLOWS FOR THE CORRECT PREDICTION OF REACTION PRODUCTS SEEN IN EXPERIMENT.



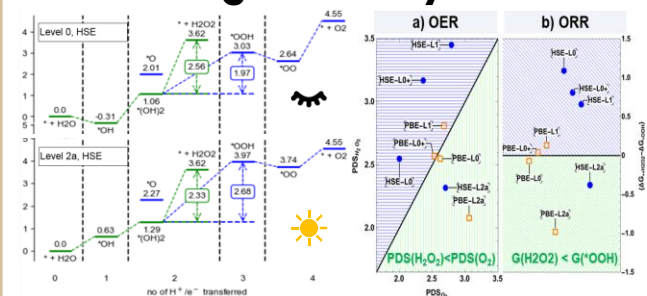
*proton-coupled electron transfer

The protocol

$$G^* = E_0 + G(T, 0 \rightarrow T) + E_{EXC}$$



Resulting thermodynamics



Excitation energy E_{EXC}

Level 0: No excitation

Level 0+: Electronic band gap

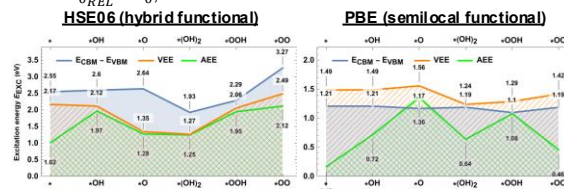
$$E_g = E_{CBM} - E_{VBM}$$

Level 1: Vertical excitation energy

(approx. optical band gap), $VEE = E_{0SP}^* - E_0$, SP – single point

Level 2: Adiabatic excitation energy

$$AEE = E_{0REL} - E_0, REL - \text{excited state relaxation}$$



Takeaways

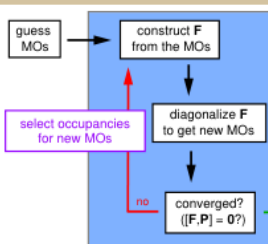
- PBE produces volatile, unphysical excitation energies. A hybrid functional should be used whenever possible
- The protocol at Level 2a predicts the correct majority product (H₂O₂) in both directions (H₂O oxidation OER, O₂ reduction ORR)
- A minimal model was used. Extensions are planned for more realism:
 - environment interaction (solvent)
 - photoelectro catalysis with constant-potential approach (grand canonical kinetics)

ΔSCF Method

Occupancy selection:

SCF (regular DFT): lowest-energy MOs are chosen as occupied according to the *aufbau* principle

ΔSCF: VBM → CBM excitation is applied



Planewave DFT, HSE06 functional

No midgap excitations

